# The *caa*<sub>3</sub> Terminal Oxidase of *Rhodothermus marinus* Lacking the Key Glutamate of the D-Channel Is a Proton Pump<sup>†</sup>

Manuela M. Pereira,‡ Marina L. Verkhovskaya,§ Miguel Teixeira,‡ and Michael I. Verkhovsky\*,§

Helsinki Bioenergetics Group, Department of Medical Chemistry, Institute of Biomedical Sciences and Biocentrum Helsinki, P.O. Box 8 (Siltavuorenpenger 10A), University of Helsinki, FIN-00014 Helsinki, Finland, and Instituto de Tecnologia Química e Biológica, Universidade Nova de Lisboa, Rua da Quinta Grande 6, 2780-156 Oeiras, Portugal

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ABSTRACT: The thermohalophilic bacterium *Rhodothermus marinus* expresses a *caa*<sub>3</sub>-type dioxygen reductase as one of its terminal oxidases. The subunit I amino acid sequence shows the presence of all the essential residues of the D- and K-proton channels, defined in most heme-copper oxidases, with the exception of the key glutamate residue located in the middle of the membrane dielectric (E278 in *Paracoccus denitrificans*). On the basis of homology modeling studies, a tyrosine residue (Y256, *R. marinus* numbering) has been proposed to act as a functional substitute [Pereira, M. M., Santana, M., Soares, C. M., Mendes, J., Carita, J. N., Fernandes, A. S., Saraste, M., Carrondo, M. A., and Teixeira, M. (1999) *Biochim. Biophys. Acta 1413*, 1–13]. Here, *R. marinus caa*<sub>3</sub> oxidase was reconstituted in liposomes and shown to operate as a proton pump, translocating protons from the cytoplasmic side of the bacterial inner membrane to the periplasmatic space with a stoichiometry of 1H<sup>+</sup>/e<sup>-</sup>, as in the case in heme-copper oxidases that contain the glutamate residue. Possible mechanisms of proton transfer in the D-channel with the participation of the tyrosine residue are discussed. The observation that the tyrosine residue is conserved in several other members of the heme-copper oxidase superfamily suggests a common alternative mode of action for the D-channel.

The last step of membrane-bound aerobic electron transfer chains, oxygen reduction, is performed by oxygen reductases, most of which belong to the superfamily of heme-copper oxidases (1-3). These enzymes are characterized by the presence of a low-spin heme and a binuclear center formed from a high-spin heme and a copper ion in subunit I. Most of them have been shown to operate as proton pumps (4): during the reduction of one dioxygen molecule to two water molecules, four protons are consumed from the cytoplasmatic or matrix space, and an additional four protons are translocated across the inner bacterial or mitochondrial membrane, contributing to an electrochemical proton gradient, which is then consumed for ATP synthesis.

Site-directed mutagenesis (5, 6) together with extensive sequence analysis and structure prediction led to the proposal that the enzyme contains two proton-conducting channels, which allow the access of protons to the catalytic center. This proposal was further confirmed by the X-ray crystallographic analysis of the structure of the  $aa_3$  oxidases from bovine heart mitochondria (7, 8) and from the bacterium Paracoccus denitrificans (9, 10). The proton pathways were named K- and D-channels according to conserved residues in these enzymes. The K-channel, named for Lys354 (K),

also contains the residues Thr351 and Tyr280. The D-channel is composed of the residues Asp124 (D), Asn199, Asn113, Asn131, Tyr35, Ser134, Ser193, and Glu278 (all P. denitrificans numbering) (7, 9, 11-13). The glutamate residue, located in the middle of the membrane dielectric, has been found to be a key element in the delivery of protons for both water formation and translocation across the membrane. Most mutations of this residue result in an enzyme incapable of completing the catalytic cycle. The oxygen reaction is blocked at the P state (11, 12), and consequently proton pumping is not observed (13-16).

Recently, an increasing number of terminal oxidases have been reported in which the typical residues of the channels are not conserved (17-26), thus indicating that alternative pathway(s) should exist. The recently characterized caa<sub>3</sub> oxidase from R. marinus (27) is particularly interesting in this context. It is the first unequivocal example of a so-called cytochrome oxidase for which the electron donor is a small, iron—sulfur protein containing a [4Fe-4S]<sup>3+/2+</sup> center (a high potential iron-sulfur protein, HiPIP) (27-29). The amino acid sequence of caa3 oxidase subunit I showed that all the residues considered important in the D- and K-channels are conserved, with the exception of the glutamate residue at the hydrophobic end of the D-channel. Homology modeling studies showed that a tyrosine residue (Y-256, R. marinus numbering) could structurally replace the glutamate. The phenol group of this residue is very close to the spatial position of the carboxyl group of the glutamate residue in the P. denitrificans and bovine enzymes. This is a natural example of a single mutation in the conserved D-channel

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<sup>\*</sup> Corresponding author: Fax: (+358) 9 191 8296. E-mail: michael.verkhovsky@helsinki.fi.

<sup>&</sup>lt;sup>‡</sup> Universidade Nova de Lisboa.

<sup>§</sup> University of Helsinki.

residue (27), thus providing a model to test the relevance of glutamate and alternative substitutes. In this work, we show that this oxidase is indeed a proton pump and discuss the

possible role of the tyrosine residue in the functioning of the D-channel.

#### MATERIALS AND METHODS

Protein Purification. Bacterial growth, membrane preparation, and solubilization were done as described in ref 30. Protein purification followed the procedure reported in ref 27.

Protein Reconstitution and Proton Pumping Measurements. Protein reconstitution was done as described in ref 31, with some alterations. Fifty microliters of R. marinus caa<sub>3</sub> oxidase (concentration approximately 60 µM) was diluted to a volume of 2 mL with a suspension of sonicated, preformed liposomes containing 80 mg/mL azolectin, 2% (w/v) cholic acid, and 100 mM Hepes-potassium, pH 7.4, and was mixed for 30 min at 4 °C. The detergent was removed by several additions of Bio-Beads SM-2 absorbent (Bio-Rad) followed by incubation periods. Bio-Beads were added to the mixture to a concentration of 66.5 mg/mL two times, and the mixture was stirred each time for 1 h at 4 °C. Then more absorbent was added (133 mg/mL), and the mixture was stirred for 2 h more also at 4 °C. Finally, two other additions (266 mg/mL each) were made to the mixture followed by 1 h stirring at room temperature. The proteoliposomes were separated from the Bio-Beads, diluted with 100 mM KCl, and pelleted by centrifugation at 144000g for 2 h to remove the external buffer. Respiratory control was determined as in ref 11. Proton pumping measurements were performed in the apparatus described in ref 32 by the O<sub>2</sub> pulse method (5, 6).

## RESULTS AND DISCUSSION

The caa<sub>3</sub> oxidase from R. marinus reconstituted into liposomes showed a respiratory control ratio of 5. The activity of the enzyme incubated with lipids prior to removal of detergent was 314 e<sup>-</sup>/s at 25 °C. The capability of the reconstituted caa3 oxidase to pump protons across the liposome membrane was investigated using the O2 pulse technique (5, 6). Figure 1 shows the acidification of the external medium when caa3 activity is initiated by the addition of air-equilibrated water. An apparent stoichiometry of 1.5 H<sup>+</sup>/e<sup>-</sup> was observed. Taking into account that 0.5 H<sup>+</sup>/ e is released upon ascorbate oxidation, a stoichiometry of proton translocation from the internal to the external aqueous phase of 1 H<sup>+</sup>/e<sup>-</sup> was calculated (protons for H<sub>2</sub>O formation are consumed inside the vesicles and do not participate in fast pH changes outside). This demonstrates exactly that like most members of the heme-copper oxidase family R. marinus caa3 operates as a proton pump.

The ability of R.  $marinus\ caa_3$  to pump protons shows clearly the presence of a functional substitute of the D-channel glutamate. The input part of this channel from the aspartate residue (D124 in P. denitrificans and D113 in R. marinus) to the serine residue (S193 in P. denitrificans and S175 in R. marinus) is shown to be very similar in both enzymes (27). In Figure 2, we present a detailed representation of the heme  $a_3$ -Cu<sub>B</sub> center and the amino acid residues of helix VI in the vicinity of the glutamate position from P.

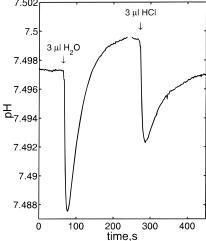


FIGURE 1: Proton translocation in  $caa_3$  oxidase from *Rhodothermus marinus*. Changes in pH of a proteoliposome suspension containing 100 mM KCl, 5 mM potassium ascorbate, 5  $\mu$ M TMPD, and 1  $\mu$ M valinomycin were measured. The anaerobic proteoliposome suspension was pulsed with air-saturated water at 25 °C. The acidification of the medium was calibrated by injection of anaerobic 1 mM HCl. An apparent stoichiometry of 1.5 H<sup>+</sup>/e<sup>-</sup> was observed, which corresponds to 1 H<sup>+</sup>/e<sup>-</sup> translocated by the  $caa_3$  oxidase across the membrane.

denitrificans oxidase (A) and the corresponding region of the previously published model of *R. marinus* enzyme (B). For the oxidases, which contain the glutamate residue, proton connectivity between the serine residue and the binuclear center has been proposed to be achieved through a network of water molecules, on the basis of the 3D structure, FTIR measurements, and theoretical calculations (14). The glutamate residue has been proposed to be a proton shuttle, which can take up, even if only transiently, two distinct conformations ("in" and "out"), related by a 180° rotation of the carboxylate group (33). Quite interestingly, in R. marinus, the spatial positions equivalent to the in and out conformations of the glutamate side chain are occupied by the hydroxyl groups of a tyrosine residue and a serine residue (Y256 and S257), respectively. While a small energy barrier of ~3 kcal/mol has been estimated for the movement of the glutamate residue (33), the phenol group of the tyrosine would be expected to present a much higher activation energy if it had to move in a similar way. In fact, taking into account the size of the tyrosine side chain and the conformational constrictions imposed by the other residues of the helix, it seems highly improbable that the tyrosine side chain could adopt different conformations.

It remains to be clarified what the actual role of the tyrosine residue and the serine residue is. They may be necessary just to maintain a well-ordered chain of water molecules in the cavity, thus enabling proton transfer without being directly involved. Alternatively, one or both might be directly involved as part of the proton wire itself, where the OH groups would function just like water molecules in a hopping mechanism. It should be mentioned that the OH group of the tyrosine may be able to adopt two opposite conformations (Mendes, J., and Soares, C. M., unpublished data), facilitating proton transfer between Ser173 of the D-channel and the binuclear center. The two alternative orientations of the hydroxyl group may be essential for the reorientation of the proton channel elements after the delivery

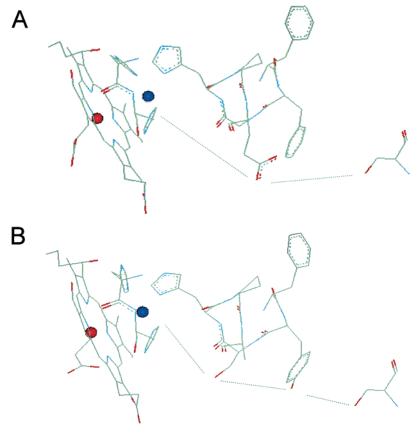


FIGURE 2: Heme  $a_3$ -Cu<sub>B</sub> centers and amino acid residues of helix VI, FFGHPE (residues 273–278) from *P. denitrificans aa*<sub>3</sub> oxidase (A) (PDB code 1AR1) and FYSHPA (residues 255–260) from *R. marinus caa*<sub>3</sub> oxidase (B) (27).

of each proton, as required by the directionality of the process.

It has also been proposed that the glutamate residue is sensitive to the oxidation state of the catalytic center and is able to be deprotonated. In fact, although theoretical calculations have led to the conclusion that the glutamate remains protonated under all equilibrium conditions, the most likely situation is that it is transiently deprotonated, thus acting as an effective proton shuttle. The pK of a glutamate residue, in solution, is 4, but large shifts may occur, depending on the amino acid environment. For example, the essential residues Asp85, Asp86, and Glu204, present in the proton channel of bacteriorhodopsin from Halobacterium halobium, show increases in pK values of 3-4 units (34). A similar situation has been proposed for the glutamate in terminal oxidases, since FTIR spectra indicate that this residue is protonated at pH values as high as 9.8 (35). A tyrosine residue has a pK of 10.5 in solution, but a higher pK value could be expected in a hydrophobic protein environment, such as the vicinity of the glutamate residue in the D-channel. Thus, it is highly unlikely that the tyrosine residue becomes deprotonated in the case of R. marinus caa3 oxidase.

Whatever the actual molecular mechanism will be, it is clear that no fully-conserved proton pathways exist in oxygen reductases of the heme-copper superfamily. While the mitochondrial-like oxidases appear to employ a carboxylate as a proton shuttle, as proposed for bacteriorhodopsin (34), this cannot be the case in *R. marinus*.

There are several enzymes in different species, which most probably have a similar mechanism of proton transport in the D-channel to the *R. marinus* oxidase. A protein database

	270	285
R.marinus caa,	FFWFYSHPAVYIN	MILP
<i>A.aeolicus</i> COXA1	IFWFYSHPVVYV	QVLP
An.sp aa,	MFWFYSHPAVYIN	MILP
S.sp aa,	LFWFYSHPAVYIN	4ILP
Sn. vulcanus aa,	LFWFYSHPAVYL1	4ILP
T.thermophilus caa,	FFWFYSHPTVYVM	1LLP
D. radiodurans	FFWFYSHPAVYVN	4LLP
S.acidocaldarius aa,	ILWFYGHPVVYY	
B.japonicum cbb <sub>3</sub>	FQWWYGHNAVGF	LTA
Ba. stearothermophilus b(o/a).	LFWYFGHPLVYFV	√LLP
N. pharaonis ba <sub>3</sub>	LIWYFGHAVVYFW	VLMP
P. denitrificans aa,	ILWFFGHPEVYMI	JILP

FIGURE 3: Alignment of subunit I amino acid sequences from oxidases containing the YS motif and only the Y residue. Sequence of the  $aa_3$  oxidase from P. denitrificans was included for comparison, as an example of an enzyme containing the glutamate residue. Genus abbreviations: R., Rhodothermus; A., Aquifex (acc C70488); An., Anabaena (Anabaena sp. strain PCC 7937, acc Z98264); S., Synechoscystis (Synechoscystis sp. strain PCC 6803, acc Q06473/P73261); Sy., Synechococcus (acc P50676); T., Thermus (acc P98005); D., Deinococcus (Preliminary sequence data was obtained from The Institute for Genomic Research website at http://www.tigr.org); Su., Sulfolobus (acc P98004); B., Bradyrhizobium (acc Q03073); Ba., Bacillus (acc O82837); N., Natronobacterium (acc CAA71525); and P., Paracoccus (acc P08305). Numbering according to P. denitrificans sequence.

search already shows six other enzymes that have the similar YS motif as in *R. marinus* (Figure 3). In evolutionary terms, these oxidases have so far been found in rather phylogenetically diverse organisms: *Cyanobacteria* (19, 21), *Aquificalles* (26), *Thermus* and *Deinococci* (20), and the *Flexibacter*, *Bacteroids*, and *Cytophaga* group (which includes

R. marinus), suggesting a parallel line of evolution. Other enzymes lacking the glutamate, such as the  $cbb_3$  oxidases (e.g., refs 18, 23, 36, and 37) and the aa<sub>3</sub> oxidase from Sulfolobus acidocaldarius (17), have a tyrosine at the same sequence position, but the serine is not present. On the other hand, the bo3 and ba3 oxidases from Bacillus stearothermophilus (38) and Natronobacterium pharaonis (24), respectively, have a tyrosine one position earlier in the sequence. Several of these enzymes have been proposed to pump protons. The caa<sub>3</sub> oxidase from Thermus thermophilus, which has a YS motif as the R. marinus oxidase, shows a stoichiometry of up to  $0.9H^+/e^-$  (39). The  $aa_3$  oxidase from S. acidocaldarius (40) also have been shown to translocate protons. The proton pumping activity of the  $cbb_3$  oxidases was assayed in whole cells from P. denitrificans (23, 41) and Rhodobacter sphaeroids (42), in which the genes for other oxidases have been knocked out. The data indicate that these enzymes may also be proton pumps, but studies with purified and reconstituted proteins are still missing.

The  $ba_3$  and  $bo_3$  cytochrome oxidases from T. thermophilus and B. stearothermophilus, respectively, do not present any of the conserved amino acid residues of the D-channel, not even the YS motif (20, 38). A stoichiometry of 0.5 H<sup>+</sup>/e<sup>-</sup> has been determined for those oxidases (43, 44), and in the case of T. thermophilus (43), it has been suggested that this pump activity is due to the functioning of a K-channel equivalent, formed by a threonine (in the place of K319, bovine numbering), a serine (in the place of T316), and a tyrosine (in the place of S255). These enzymes represent a completely different group from the mitochondrial-like oxidases, containing the glutamate residue or the YS motif.

Although the actual molecular details of the mechanism in both *R. marinus* and mitochondrial-like oxidases have yet to be fully established, a general picture emerges: the essential features are a proton channel, which allows proton conductivity through the protein from the internal to the external membrane phases, made up of a network of hydrophilic amino acid residues maintaining a chain of water molecules, and a gate element, which prevents short-circuiting. The only common elements of all heme-copper oxidases appears to be the prosthetic groups of subunit I and their ligands. Considering the prosthetic groups, only Cu<sub>B</sub> is always conserved, since the chemical nature of the heme can vary. This further strengthens the hypothesis that the actual gating is performed by the binuclear reaction center (45–47).

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## REFERENCES

- Trumpower, B. L., and Gennis, R. B. (1994) *Annu. Rev. Biochem.* 63, 675–716.
- van der Oost, J., de Boer, A. P. N., de Gier, J. W. L., Zumft, W. G., Stouthamer, A. H., and van Spanning, R. J. M. (1994) FEMS Microbiol. Lett. 121, 1–10.
- Garcia-Horsman, J. A., Barquera, B., Rumbley, J., Ma, J., and Gennis, R. B. (1994) *J. Bacteriol.* 176, 5587–5600.
- Wikström, M., and Krab, K. (1980) Curr. Top. Bioenerg. 10, 51–101.

- 5. Thomas, J. W., Puustinen, A., Alben, J. O., Gennis, R. B., and Wikström, M. (1993) *Biochemistry* 32, 10923–10928.
- 6. Garcia-Horsman, J. A., Puustinen, A., Gennis, R. B., and Wikström, M. (1995) *Biochemistry 34*, 4428–4433.
- 7. Tsukihara, T., Aoyama, H., Yamashita, E., Tomizaki, T., Yamaguchi, H., Shinzawa-Itoh, K., Nakashima, R., Yaono, R., and Yoshikawa, S. (1995) *Science* 269, 1069–1074.
- 8. Tsukihara, T., Aoyama, H., Yamashita, E., Tomizaki, T., Yamaguchi, H., Shinzawa-Itoh, K., Nakashima, R., Yaono, R., and Yoshikawa, S. (1996) *Science* 272, 1136–1144.
- 9. Iwata, S., Ostermeier, C., Ludwig, B., and Michel, H. (1995) *Nature 376*, 660–669.
- Ostermeier, C., Harrenga, A., Ermler, Y., and Michel, H. (1997) Proc. Natl. Acad. Sci. U.S.A. 94, 10547–10553.
- Verkhovskaya, M. L., Garcìa-Horsman, A., Puustinen, A., Rigaud, J. L., Morgan, J. E., Verkhovsky, M. I., and Wikström, M. (1997) Proc. Natl. Acad. Sci. U.S.A. 94, 10128-10131.
- Karpefors, M., Ädelroth, P., Aagaard, A., Sigurdson, H., Ek, M. S., and Brzezinski, P. (1998) *Biochim. Biophys. Acta* 1365, 159–169.
- 13. Gennis, R. B. (1998) Biochim. Biophys. Acta 1365, 241-248.
- Riistama, S., Hummer, G., Puustinen, A., Dyer, R. B., Woodruff, W. H., and Wikström, M. (1997) FEBS Lett. 414, 275–280.
- 15. Wikström, M. (1998) Biochim. Biophys. Acta 1365, 185-192.
- Mills, D. A., and Ferguson-Miller, S. (1998) *Biochim. Biophys. Acta* 1365, 46–52.
- 17. Lübben M., Kolmerer B., and Saraste M. (1992) *EMBO J.* 11, 805–812.
- Preisig, O., Anthamatten, D., and Hennecke, H. (1993) *Proc. Natl. Acad. Sci. U.S.A.* 90, 3309–3313.
- 19. Sone, N., Tano, H., and Ishizuka, M. (1993) *Biochim. Biophys. Acta 1183*, 130–138.
- Mather, M. W., Springer, P., Hensel, S., Buse, G., and Fee, J. A. (1993) J. Biol. Chem. 268, 5395-5408.
- 21. Alge, D., Schmetterer, G., and Peschek, G. A. (1994) *Gene* 138, 127–132.
- Keightley J. A., Zimmermann B. H., Mather M. W., Springer P., Pastuszyn A., Lawrence D. M., and Fee J. A. (1995) *J. Biol. Chem.* 270, 20345–20358.
- 23. Raitio, M., and Wikström, M. (1994) *Biochim. Biophys. Acta* 1186, 100–106.
- Mattar S., and Engelhard M. (1997) Eur. J. Biochem. 250, 332–341.
- Purschke W. G., Schmidt C. L., Petersen A., and Schäfer G. (1997) J. Bacteriol. 179, 1344–1353.
- Deckert, G., Warren P. V., Gaasterland T., Young W. G., Lenox A. L., Graham D. E., Overbeek, R., Snead M. A., Keller M., Aujay M., Huber R., Feldman R. A., Short J. M., Olsen G. J., and Swanson R. V. (1998) *Nature* 392, 353–358.
- 27. Pereira, M. M., Santana, M., Soares, C. M., Mendes, J., Carita, J. N., Fernandes, A. S., Saraste, M., Carrondo, M. A., and Teixeira, M. (1999) *Biochim. Biophys. Acta* 1413, 1–13.
- Pereira, M. M., Carita, J. N., and Teixeira, M. (1999) *Biochemistry* 38, 1276–1286.
- Pereira, M. M., Antunes, A. M., Nunes, O. C., Costa, M. S., and Teixeira, M. (1994) FEBS Lett. 352, 327–330.
- 30. Pereira, M. M., Carita, J. N., and Teixeira, M. (1999) *Biochemistry* 38, 1268–1275.
- Jasaitis, A., Verkhovsky, M. I., Morgan, J. E., Verkhovskaya, M. L., and Wikström, M. (1999) *Biochemistry* 38, 2697–2706.
- 32. Verkhovsky, M. I., Belevich, N., Morgan, J. E., and Wikström, M. (1999) *Biochim. Biophys. Acta 1412*, 184–189.
- 33. Pomès, R., Hummer, G., and Wikström, M. (1998) *Biochim. Biophys. Acta* 1365, 255–260.
- 34. Lanyi, J. K. (1997) J. Biol. Chem. 272, 31209-31212.
- 35. Lübben, M., Prutsch, A., Mamat, B., and Gerwert, K. (1999) *Biochemistry* 38, 2048–2056.
- 36. Garcia-Horsman, J. A., Berry, E., Shapleigh, J. P., Alben, J. O., and Gennis, R. B. (1994) *Biochemistry 33*, 3113–3119.
- Gray, K., Grooms, M., Myllykallio, H., Moomaw, C., Slaughter, C., and Daldal, F. (1994) *Biochemistry* 33, 3120–3127.
- 38. Nikaido, K., Noguchi, S., Sakamoto, J., and Sone, N. (1998) *Biochim. Biophys. Acta* 1397, 262–267.

- 39. Hon-nami, K., and Oshima, T. (1984) *Biochemistry*, 23, 454–460.
- Gleissner, M., Kaiser, U., Antonopoulos, E., and Schäfer, G. (1997) J. Biol. Chem. 272, 8417–8426.
- 41. de Gier, J. W. L, Lübben, M., Reijnders, W. N. M., Tipker, C. A., Slotboom, D. J., van Spanning, R. J. M., Stouthamer, A. H., and van der Oost, J. (1994) *Mol. Microbiol.* 13, 183–196.
- 42. Toledo-Cuevas, M., Barquera, B., Gennis, R. B., Wikström, M., and Garcìa-Horsman, J. A. (1998) *Biochim. Biophys. Acta* 1365, 421–434.
- 43. Kannt, A., Soulimane, T., Buse, G., Becker, A., Bamberg, E., and Michel, H. (1998) *FEBS Lett.* 434, 17–22.
- 44. Nikaido, K., Sakamoto, J., Noguchi, S., and Sone, N. (2000) *Biochim. Biophys. Acta*, 1456, 35–44.
- 45. Wikström, M. (1989) Nature 337, 776-779.
- 46. Morgan, J. E., Verkhovsky, M. I., and Wikström, M. (1994) *J. Bioenerg. Biomembr.* 26, 599–608.
- 47. Wikström, M., Morgan, J. E., and Verkhovsky, M. I. (1998) J. Bioenerg. Biomembr. 30, 139–145.

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